

PHYSICAL SCIENCES GRADE 10 2020 COMMON TESTS AND EXAMINATION SCOPE

MARCH CONTROL TESTS			
PAPER	TOPICS	MARK	DURATION
Paper 1	Waves, Sound & Light: Transverse pulses on a string or spring, Transverse & Longitudinal Waves, Sound and Electromagnetic Radiation.	50	1 hour
Paper 2	Matter & Materials: Revision of Matter & Classification, State of Matter Kinetic Molecular Theory, The Atom: Basic building blocks of Matter, Periodic Table, Chemical Bonding; Particles substances are made of (cf. molecular and ionic substances)	50	1 hour
	Last date for this term work is 06 March	2020	
	MID-YEAR EXAMS		
NB: Mid-year exam	s for all grades assess all the first and second term's work.		
PAPER	TOPICS	MARKS	DURATION
Paper 1	Waves, Sound & Light: Transverse pulses on a string or spring, Transverse & Longitudinal Waves, Sound, Electromagnetic Radiation. Electricity & Magnetism: Magnetism, Electrostatics, Electric Circuits.	100	2 hours
Paper 2	Matter & Materials: Matter & Classification, KMT, The Atom: Basic building blocks of Matter, Periodic Table, Chemical Bonding, Particles Substances are made of. Chemical Change: Physical & Chemical Change, Representing Chemical Change; Reactions in aqueous solution	100	2 hours
	SEPTEMBER CONTROL TESTS		1
NB: September con	trol tests for grade 10 and 11 will assess all the 3 term's work or term	1, 2 and term 3 v	vork.
PAPER	TOPICS	MARK	DURATION
Paper 1	Mechanics: 50% Vectors & Scalars, Motion in one dimension, Instantaneous Speed, Velocity & Equations of Motion. Energy (types) and conservation of mechanical energy Waves, Sound & Light: 20% Transverse pulses on a string or spring, Transverse & Longitudinal Waves, Sound and Electromagnetic Radiation. Electricity & Magnetism: 30% Magnetism, Electrostatics, Electric Circuits.	150	3 hours
Paper 2	Matter & Materials: 50% Matter & Classification, KMT, The Atom: Basic building blocks of Matter, Periodic Table, Chemical Bonding, Particles Substances are	150	3 hour

made of.	
Chemical Change: 50% Physical & Chemical Change, Representing Chemical Change; Reactions in aqueous solution, Quantitative aspects of chemical change.	

FINAL EXAMINATION: GRADE 10

FORMAT OF QUESTION PAPERS

Paper 1: Physics 2 hours	Paper 2: Chemistry 2 hours
SECTION A:	SECTION A:
Multiple-choice questions	Multiple-choice questions
SECTION B: Conceptual questions assessing all themes	SECTION B: Conceptual questions assessing all themes
Total: 150 marks	Total: 150 marks

MARK ALLOCATION PER KNOWLEDGE AREA: FINAL EXAMINATIONS GR 10 PAPER 1

Knowledge Area	Theme	Marks	
	Vectors & Scalars		
Mechanics	Motion in one dimension	75	
(50%)	Instantaneous Speed, Velocity & Equations of Motion	75	
	Energy		
	Transverse pulses on a string or spring		
Waves, sound and light	Transverse and Longitudinal Waves	40	
(±26.7%)	Sound	40	
	Electromagnetic Radiation; Indigenous Knowledge Syst		
Electricity and magnetism	Magnetism		
Electricity and magnetism	Electrostatics	35	
(±23,3%)	Electric Circuits		
	TOTAL	150	

PAPER 2

Knowledge Area	Theme	Marks	
	Matter & Classification; Kinetic Molecular Theory		
Mattananalmataniala	The Atom: Basic building blocks of Matter		
Matter and materials $(\pm 46,7\%)$	Periodic Lable		
	Chemical Bonding		
	Particles Substances are made of		
	Physical & Chemical Change		
Chemical change	Representing Chemical Change	60	
(40%)	Reactions in aqueous solution	80	
	Quantitative aspects of Chemical Change		
Chemical systems (±13.3%)	The Hydrosphere	20	
. ,	TOTAL	150	

COGNITIVE LEVELS

Cognitive level description	Weigh	Weighting %		
Cognitive level description	Paper 1 Paper 2			
Remembering	15	15		
Understanding	35	40		
Applying and Analysing	40	35		
Evaluating	10	10		